# Competition between efficiency and selectivity in ring assembly

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	Recent work on SAS-6	5 rin	gs		
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## HS-AFM imaging with photothermal off-resonance tapping (PORT)

Os	2.5s	5.1s	7.7s	10.2s
12.8s	15.3s	17.9s	20.4s	23s
25.6s 50 nm	28.1s	30.7s	33.2s	35.8s

HS-PORT is sufficiently gentle such that SAS-6 ring assembly can be monitored with high temporal and spatial resolution.

[Nievergelt, Banterle et al. Nature Nanotech 2018]

















- Standard approach for reversible assembly of clusters and polymers
- Go back to Smoluchowski 1916, Becker and Döring 1935, Kolmogorov 1941
- Widely used not only in biophysics and soft matter physics, but also in astrophysics, environmental physics, geosciences, etc



### CF-equations for complexes of maximal size N

$$\dot{c}_{j}(t) = \frac{1}{2} \sum_{k=1}^{j-1} a_{j-k,k} c_{j-k} c_{k} - \sum_{k=1}^{N-j} a_{j,k} c_{j} c_{k} - \frac{1}{2} \sum_{k=1}^{j-1} b_{j-k,k} c_{j} + \sum_{k=1}^{N-j} b_{j,k} c_{j+k}$$

- System of ordinary differential equations for the concentrations  $c_j(t)$  of the clusters of size j as a function of time t

- Mass action kinetics; because of the binary coagulation terms, the equations are non-linear
- · Gain and loss terms have plus and minus signs, respectively
- The factors ½ avoid overcounting (1+4->5 is the same process as 4+1->5)
- In 3D association rates a are measured in 1/Ms or m<sup>3</sup>/s; in 2D they are measured in m<sup>2</sup>/s; dissociation rates b are measured in 1/s
- The CF-equations obey mass conservation:  $\sum_{j=1} j \dot{c}_j(t) = 0$
- SAS-6 rings: N=9; for macroscopic systems, one often uses N->infinity.











#### Binding as 2-step process - continued

A convenient way to rewrite our results:

$$\frac{1}{k_f} = \frac{1}{k_+} + \frac{1}{k_{on}}\frac{k_-}{k_+}, \ \frac{1}{k_r} = \frac{1}{k_-}\frac{k_{on}}{k_{off}} + \frac{1}{k_{off}}$$

Diffusion rates from solving the diffusion equation (Smoluchowski rates):

$$k_{+} = 4\pi Ds, \ k_{-} = \frac{3D}{s^{2}}$$

The diffusion constant D is universal due to the Stokes-Einstein relation (around  $(10 \text{ um})^2$ /s with s=4 nm). A typical value for the Smoluchowski association rate therefore is  $6 \times 10^9$  1/Ms. This is reduced by 4-5 orders of magnitude if one considers orientational degrees of freedom (Solc and Stockmayer).













## Summary

- SAS-6 dimers forms 9-fold rings that can be observed with HS-AFM on mica surfaces
- The surface shifts the binding equilibrium up by four orders of magnitude compared to bulk, by concentrating the molecules
- This explains why daugther centrioles form on the surface of mother centrioles
- Due to the fast kinetics, the assembly curves can be fitted by the coagulation-fragmentation (CF) model
- Frequent dissociation avoids kinetic traps and leads to successful assembly, similar to the situation with viruses
- The comparison between reaction kinetics (CF-equations) and spatial simulations (Brownian Dynamics) showed the limits of the kinetic approach: spatial effects would be more important if interaction energies were higher
- Like for viruses, malformed structures can form (7-, 8-, 10-rings, squeelices)
- The simulations reveal the intrinsic challenge to balance efficiency and selectivity and suggests that additional molecular mechanisms must exist in the cell to select the 9-fold rings (similar to scaffolding proteins like P22 for viruses)



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