The rough journey of Brownian motion: from pollen particles, to Avogadro number, to stock markets


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Even inside quartz with trapped water!

## Molecular motion



James Maxwell (1859), Ludwig Boltzmann (1871): kinetic theory.


Motivation: explain the ideal gas law $P V \propto T$. Important link to Brownian motion...

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- Many were sceptical of kinetic theory: seen only as explanatory.


## Brownian motion explained



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bombardment by tiny particles causes observable movement.


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Idea not new, but Einstein and Smoluchowski made it quantitative.

## Random walk

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But on average, 10 steps sideways.

## 20 ants moving together:



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- Therefore $x_{n+1}^{2}$ is $x_{n}^{2}+1$ on average.

Much slower than directed movement.


## Random walk in 2D



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- The average of $\frac{1}{2} m v^{2}$ is $T / N$ - thermal equilibrium.
- Using symmetry and $m$ small, deduce at time $t$ :

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\text { average of } x^{2}=4 t \frac{T}{r N}
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Jean Perrin (1909):
measured Avogadro's number as $6.4 \times 10^{23}$
(true value is $6.02 \times 10^{23}$ )
$\leadsto$ Nobel Prize 1926.


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Two ways to understand randomness:

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In classical mechanics, there is no intrinsic randomness (cf. quantum mechanics).

Brownian motion in gas is deterministic.
Nonetheless we gain a lot by treating it as random.

## Mathematics of Brownian motion

Norbert Wiener (1928): mathematical construction. Then Paul Lévy, Kiyosi Itô, ...
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## Mathematics of Brownian motion

Take 2D Brownian motion and zoom in. It stays a Brownian motion!


## Stock prices

Brownian motion models randomness in finance.


Louis Bachelier (1900): "The Theory of Speculation": stock prices $S_{t}$ behaves like Brownian motion

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- (Revival of Bachelier model in 2020 with negative oil prices!)


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- What? Why $\sigma^{2} t / 2$ ?
- Can be guessed: average of $e^{\sigma B_{t}}$ is $e^{\sigma^{2} t / 2}$.
- Brownian motion does not respect rules of calculus.


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- Connections between Harmonic Analysis, Stochastics, and PDEs are being explored.


## Thank you for listening!

## Happy Summer Solstice

