

# Mathematical Model of Heat Damage of Cells

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# Motivation

MATH 454 Take-home Exam:

How long can a person hold a coffee stirrer?



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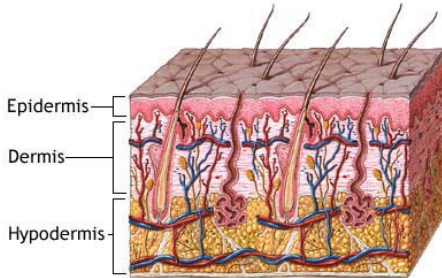
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# Motivation

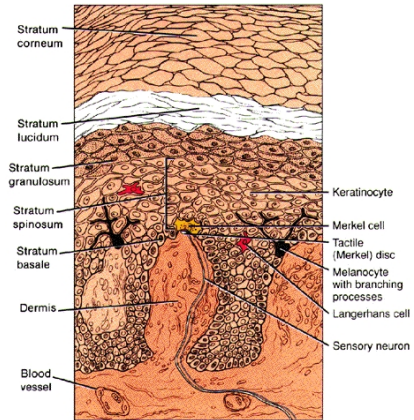
Burn injuries are a typical case of heat damage to living cells:



# Skin Anatomy



ADAM.



# Degree of Burn [Diller]

## First Degree Burns

- Moderate vasodilatation, increased permeability of endothelium, reddening of the tissue.

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- Capillary damage;
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### Second Degree Burns

- Capillary damage;
- High vasodilatation, large gaps in endothelium;
- Fluid loss, drop in plasma volume.

### Third Degree Burns

- Burn of all epidermal elements;
- Local blood vessels obliterated.

# Epidermal Wound Healing, Heat Damage of Cells

## Epidermal Wound Healing [Murray]

- Diffusion of living cells into dead regions;
- Mitotic generation of new cells;
- First order kinetic, traveling waves.

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## Thermal Damage of Cells [Henriques]

- Thermal denaturation (unfolding) of enzymes;
- Alterations in cell metabolic processes;
- Nonprotein alterations in cells as changes in ion concentrations, break up of lipid membrane, etc.

# Key Factors in Burn Injury

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- Thermal denaturation (unfolding) of enzymes;
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- Heat flow in the non-homogeneous skin environment.

# Experimental Data [Henriques & Mortiz]

## Standard hyperthermia reference data

- Multiple skin burns of human and porcine skin ( $44^{\circ}$ ,  $100^{\circ}$ );
- Single and repeated exposures of a fixed time;
- Dermal reactions categorized after a certain time;
- Duplicate exposures in borderline cases.

## Experimental Data [Henriques & Mortiz]

### Major observations

- Very small quantitative difference in the susceptibility of **human and porcine** epidermis to thermal injury;
- **Optimal thermal milieu** of the epidermal cells lies within a few degrees of normal physiological temperature;
- **Two phases**: heat saturation (temperature growth), stable temperature gradient;
- **Repeated exposure** does not cumulate in a simple way, recovery period influences damage;
- **Pressure** does not significantly influence thermal injury.

## Further Literature

- **Medicine** (Thermal wounds, treatment, healing; hyperthermal cancer treatment; skin aging);
- **Biomechanics and biomedical engineering** (Details of thermal injury process, heat-shock response, vascular permeability);
- **Fire protection** (Comparison of different fire protecting materials);
- **Mathematical biology** (Numerical solutions to heat flow in human skin coupled to vascular flow, different means of heat transport);
- **Biochemistry** (Alternations of metabolic processes, glycolysis);
- **Biomedical Optics** (Thermometry methods).

# Moritz and Henriques: Damage Function

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A first order chemical reaction process – **the Arrhenius equation**:

$$\frac{d\Omega}{dt} = Ae^{-G/RT}.$$

- $\Omega(t)$  – a damage function;
- $G$  – an activation energy for the reaction;
- $T(t)$  – absolute temperature of a local tissue (calculated from the heat equation at the basal layer).

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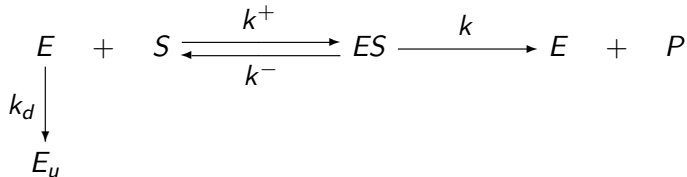
- $\Omega(t)$  – a damage function;
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The **total thermal damage** over the time interval  $t_0, t_1$

$$\Omega = \int_{t_0}^{t_1} Ae^{-G/RT(t)} dt.$$

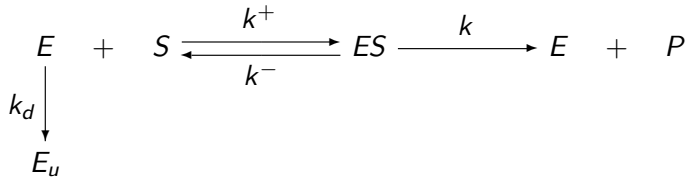
# Xu and Qian: Improved Model

Based on **Michaelis-Menten kinetics** and **transition state theory**:

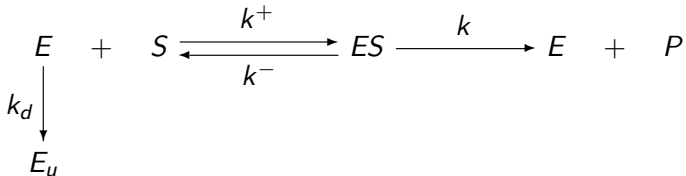


- $E$  – concentration of an enzyme;
- $S$  – concentration of a substrate;
- $ES$  – concentration of a complex;
- $P$  – concentration of a product;
- $E_u$  – concentration of an unfolded enzyme.

# Xu and Qian: Improved Model



# Xu and Qian: Improved Model



## Assumptions

- $k_d \ll k$ ;
- $k_- \ll k$ ;
- only free form of the enzyme is deactivated;
- the substrate concentration  $S$  constant during the process.

# Xu and Qian: Improved Model

Transition state theory:

$$k_d = k_{d0} \exp \left( -\frac{G_d}{RT} + \frac{G_d}{RT_0} \right) = r_0 e^{\alpha\theta}.$$

Parameters:

- $\alpha = G_d RT_0$ ;
- $\theta = (T - T_0)/T_0$ ;
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Then using Quasi-Steady State approximation

$$\Omega = \int_{t_0}^{t_1} \frac{Ae^{\alpha\theta}}{1 + Be^{-\beta\theta}} dt.$$

# Xu and Qian: Criticism

## Biological Reasons

- Only a single cell damage, parameter matching without the heat equation.
- Unlimited damage for any temperature.
- Bad match for large temperatures.
- Does not match repeated exposure experiments.

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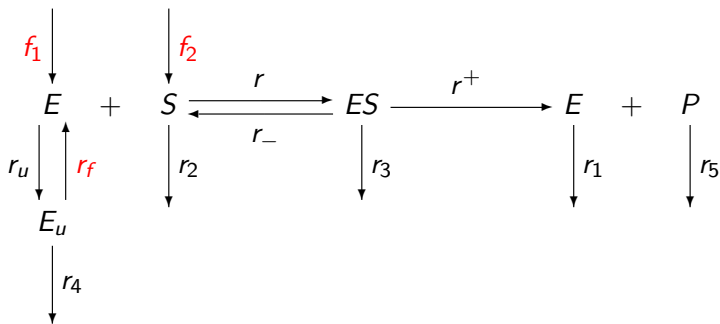
## Mathematical Reasons

- Without sources, the system goes to equilibrium.
- Quasi-steady state not reached in numerical simulations.

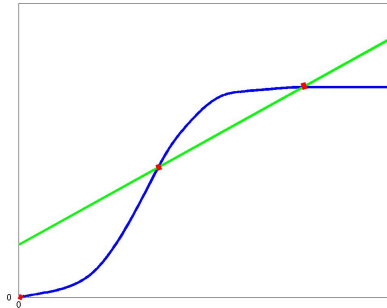
# Assumptions

- **Unfolding** and **refolding** of enzymes;
- Zero state (**dead cell**) is a stable equilibrium;
- At a physiological level there is a **nontrivial stable equilibrium**;
- **Sources of enzyme** and substrate depend on the state of the system;
- **Sources of the substrate** are due to cell **exterior** with saturation;
- 1 **Neglect** the effect of increased **permeability** of blood vessels.

# Proposed Improved Model for a Single Cell

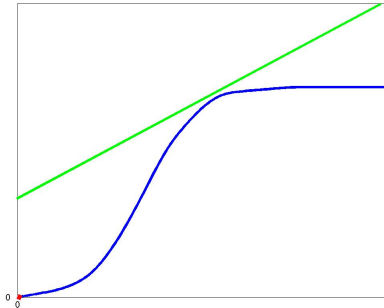


## Equilibria of the Proposed Improved Model



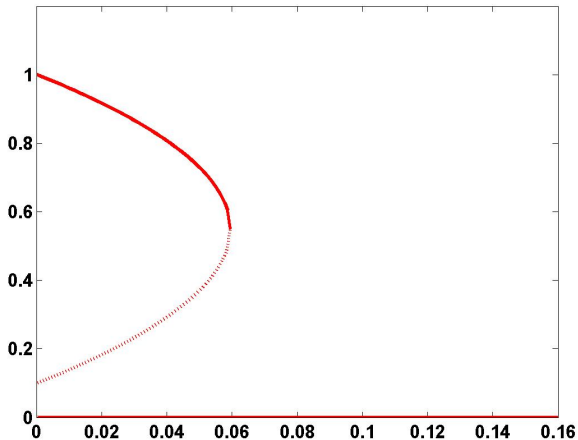
**Figure:** Two intercepts correspond to two equilibria of the system, the trivial state is an equilibrium as well.

## Equilibria of the Proposed Improved Model

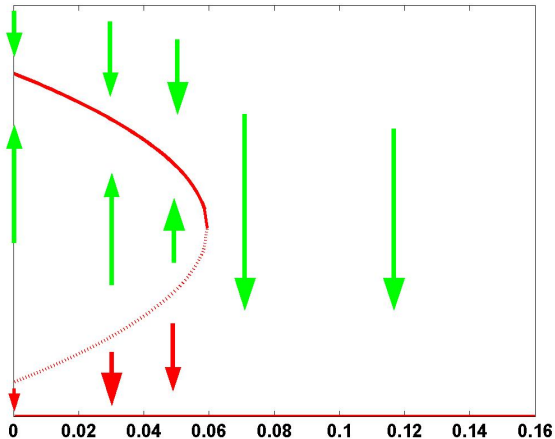


**Figure:** Under a reasonable assumption on activation energies the curves do not intersect for a larger temperature and the two equilibria will disappear via a **saddle-node bifurcation**. A **ghost of equilibria** appears close to the trivial state.

# Simplified Saddle-Node Bifurcation Model



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## Important Features

- **The attractive trivial state** for all temperatures (dead cell), exponential rate very close to zero;
- At temperatures close to physiological temperature **stable non-trivial equilibrium**;
- At temperatures very close to normal physiological temperature, the system is able to adjust, particularly thermal cell death does not happen.
- **A ghost of equilibria** (critical slow-down) closer and closer to zero with an increasing temperature;
- Possible **hysteresis** effect that explains repeated exposure data;

# Simplified Saddle-Node Bifurcation Model

## Additional Assumptions

- Biochemical processes in the cell are governed by the **transition state theory**.
- The system is governed by three principal rates:
  - the typical rate of the **main** chemical reaction,
  - the typical rate of **deactivation/degradation/melting** of various chemicals,
  - the typical rate of **denaturation** (unfolding of proteins).

# Simplified Saddle-Node Bifurcation Model

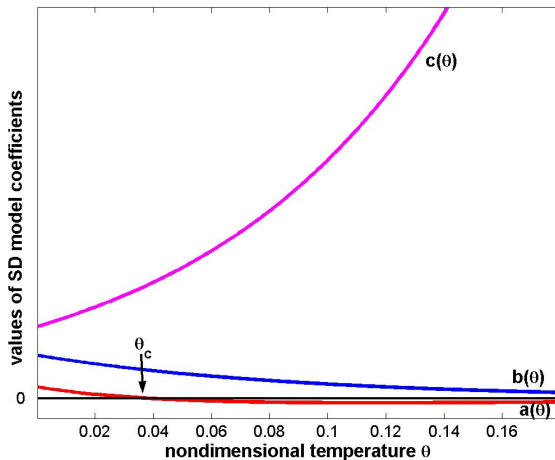
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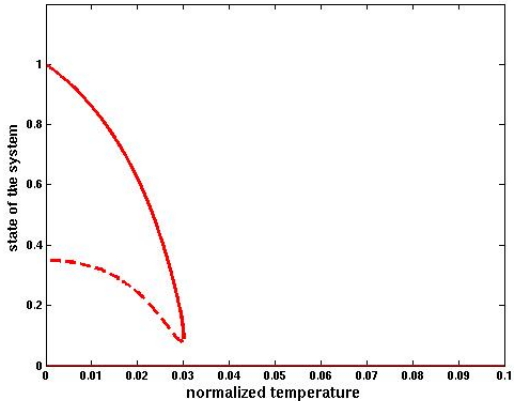
## Saddle-Node Bifurcation Model

$$\frac{dy}{dt} = f(y) = -cy ((y - b)^2 - a) .$$

## Coefficients of the Saddle-Node Model



# Saddle-Node Bifurcation Model Including Transition State Theory



## Real Setting

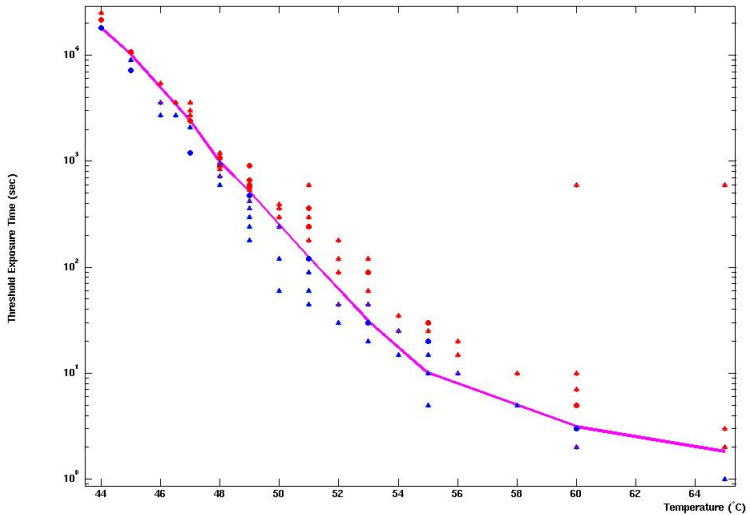
- Heat conduction modeled by a **3D heat equation**;
- Heat is also **convected** by fluid in blood vessels;
- A **Newton** boundary condition outside;
- **Nonhomogeneous** environment changing with the burn injury;
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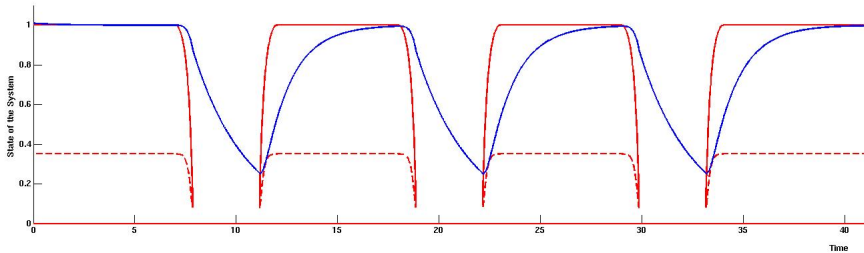
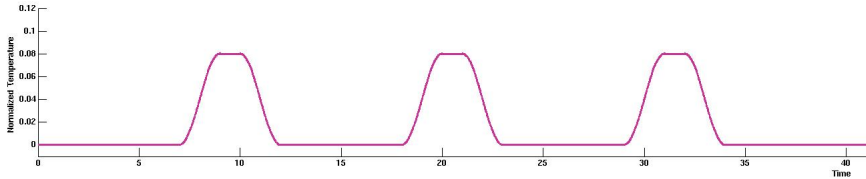
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## Simple Mathematical Model

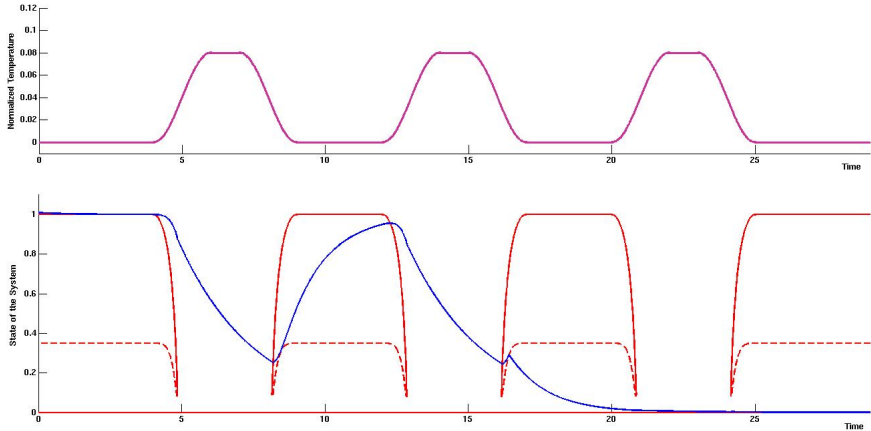
- Heat conduction modeled by a **1D heat equation**;
- **Constant** temperature boundary condition outside;
- **Homogeneous** environment;
- A homogeneous **Neumann** boundary condition ( $dT/dz = 0$ ) at the inner side.



# Repeated Exposure Results I



# Repeated Exposure Results II



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- We considered a **general** chemical kinetics **model** with sources and proved that the system **undergoes a saddle-node bifurcation** at some critical temperature slightly above the normal physiological conditions.
- We suggested **a simple phenomenological model** for the saddle-node bifurcation.
- We **matched** the available parameters to results of biomedical **experiments**.

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- Add **radiation and convection** to the heat flow;
- Consider (or derive) a more **realistic boundary condition** at the point of contact with the hot object.
- **Experimentally test the prediction** for small and large temperatures.

